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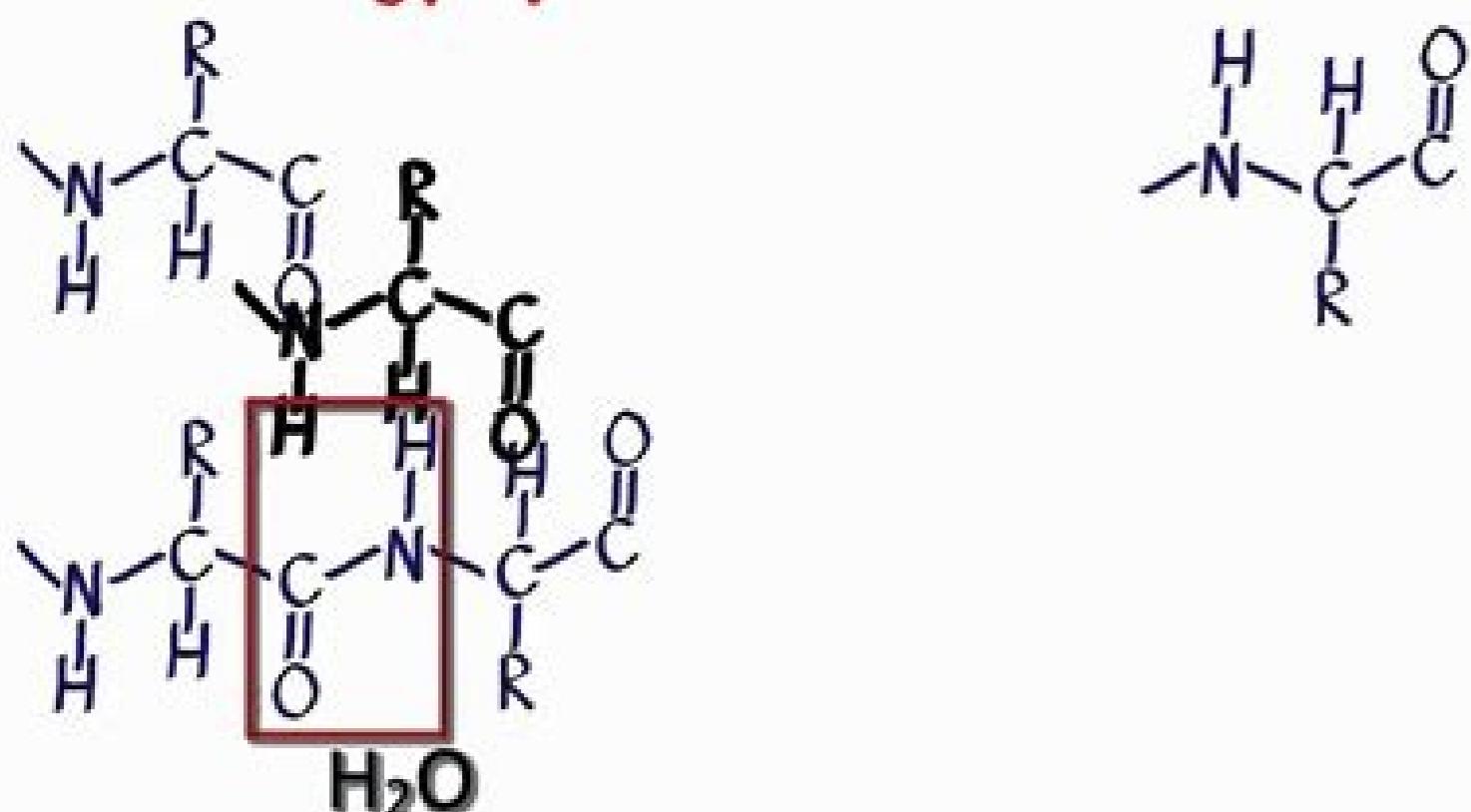


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Formation of carbonic acid reaction

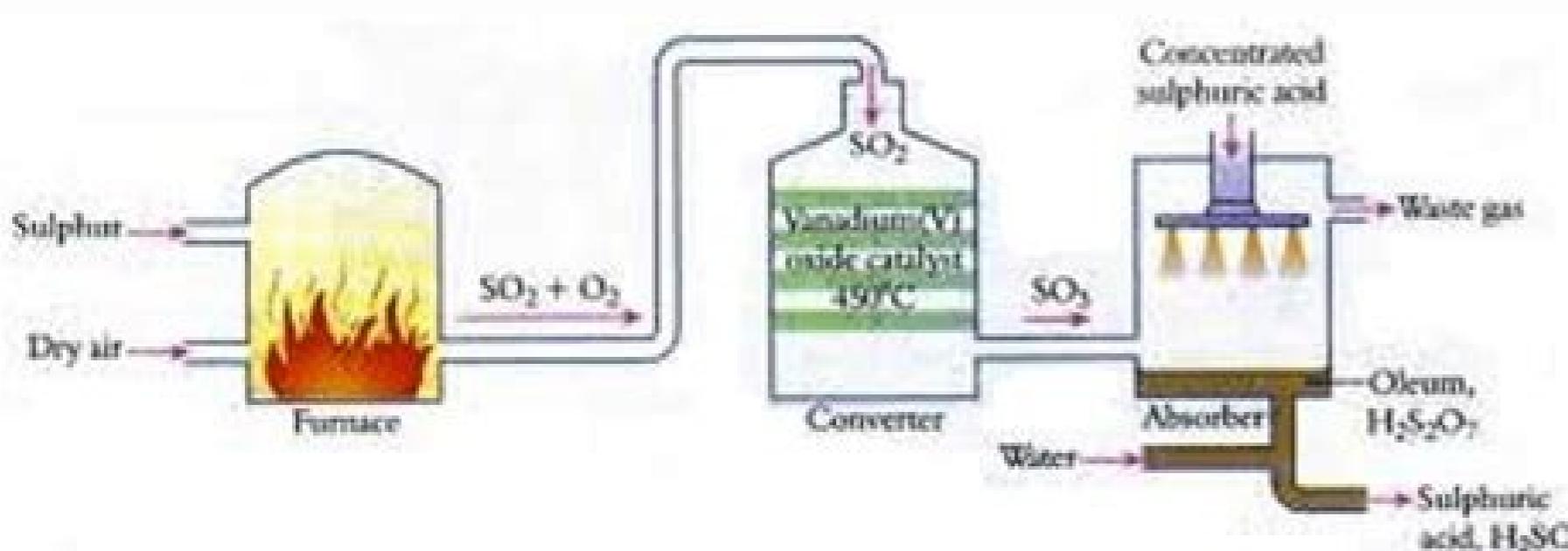


Polypeptide formation.



The diagram illustrates the formation of acid rain and its environmental impacts:

- Reaction and Transformation:** Sulfur dioxide (SO_2) and nitrogen oxides (NO_x) undergo oxidation to form sulfuric acid (H_2SO_4) and nitric acid (HNO_3).
- Cloud Interaction:** The acids react with clouds to produce H^+ ions and NO_3^- ions.
- Deposition:** The process leads to two types of deposition:
 - Dry Deposition:** Represented by a downward arrow.
 - Wet Deposition:** Represented by a blue downward arrow.
- Effects:** Both deposition pathways lead to various environmental effects, including:
 - Soil and Vegetation
 - Inland Aquatic Material
 - Human
- Emission Sources:** A red upward arrow points to the oxidation stage, indicating sources like Factories, Automobiles, Volcano, and etc.



The reaction of ammonium hydroxide and carbonic acid results in the formation of

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